

**Series 12**

**05 December 2025**

**Exercise 1: Precipitation sequence in Al-Cu**

- a) We generally observe that coherent precipitates are small (nanometric). Why?
- b) Suppose:
- 1) The elastic energy per unit volume of a flat precipitate (penny-shape) is given by  $\Delta G_{el} = \frac{3}{2} E \eta^2$ .  $E$  is the Young modulus, and  $\eta$  is the relative expansion of the precipitate versus the matrix.
  - 2) The alloy Al-4%Cu has a precipitation sequence:  $GP \rightarrow \theta'' \rightarrow \theta' \rightarrow \theta$ . The most stable precipitates increasingly lose their coherence with the matrix.
  - 3) Data:  $E = 70 \text{ GPa}$ ,  $\gamma = 0.5 \text{ J/m}^2$ , the aspect ratio  $A = \text{radius/thickness}$  is a constant equal to 5. The crystalline structure of the precipitates is represented in the following figure.

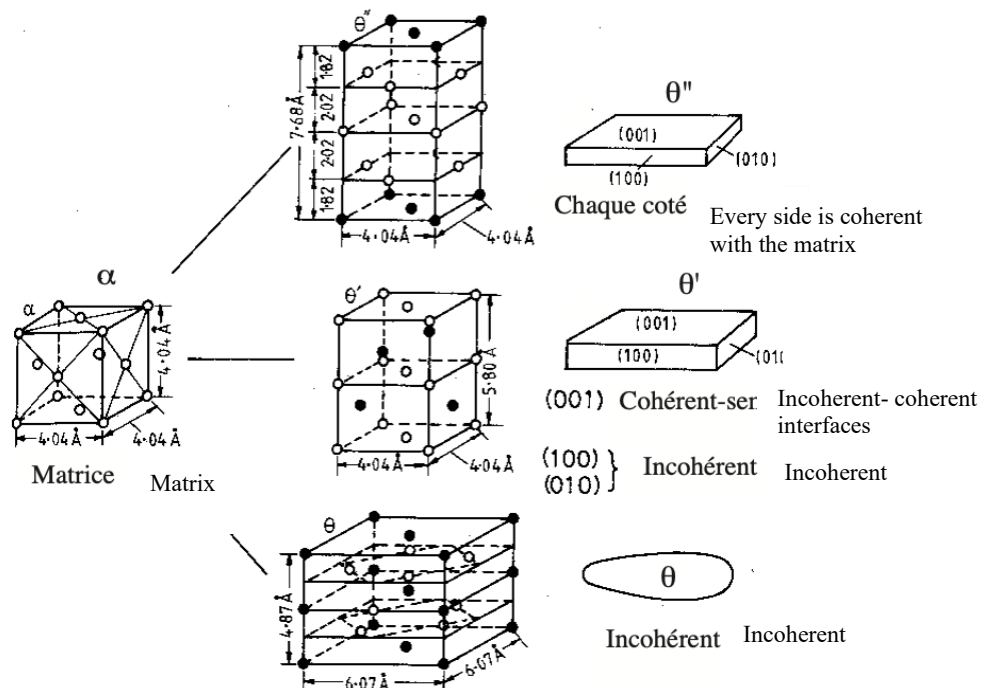


Figure 1 Representation of primitive cells of  $\theta$  phases in Al-Cu

What is the critical thickness to pass from the formation of GP zones to the formation of  $\theta''$  precipitates?

## Exercise 2: Localized nucleation

In the case of localized nucleation on a flat surface, show that:

$$R_c^{loc} = -\frac{2\gamma_{SL}}{\Delta g_V} \quad \text{and} \quad \Delta G_c^{loc} = \frac{4\pi}{3} \frac{\gamma_{SL}^3}{\Delta g_V^2} (2 - 3\cos\theta + (\cos\theta)^3)$$

where  $\Delta g_V = g_S - g_L$  is the free energy per volume unit variation during the solidification, and  $\gamma_{SL}$  is the "solid-liquid" interface energy per surface unit.